106. Synthesis of Reactive Intermediates for Natural Cyclopentane Derivatives from (-)-Quinic Acid: Preparation of 11a-Hydroxy-13-oxaprostanoic Acid

by Jean-Claude Barrière, Angèle Chiaroni, Jeanine Cléophax, Stephan D. Géro, Claude Riche and Marc Vuilhorgne

Institut de Chimie des Substances Naturelles, CNRS, 91190 Gif-sur-Yvette, France

Dedicated to the memory of Prof. Frantisek Sorm

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Summary

We describe a flexible, stereocontrolled synthesis of enantiomerically pure substituted cyclopentenes and cyclopentanes – including 11a-hydroxy-13-oxa-prostanoic acid **20** – from (–)-quinic acid *via* an acyclic precursor **6** by an intra-molecular aldolisation-dehydration reaction.

Synthetic approaches to enantiomerically pure, substituted cyclopentanes are of high interest, because they are decisive intermediates in the total synthesis of natural cyclopentane derivatives [1] such as prostaglandines [2] [3], methylenomycins [4], bredfeldines [5] and pentenomycines [6]. Since our manifold interest in the natural cyclopentane derivatives lies especially in their pharmacological activity, we needed cyclopentanes which are substituted in one proper natural configuration.

We report in this communication, a short and flexible, stereocontrolled synthesis of cyclopentenes of types 7 (3-benzyloxy-5, 5-(ethylenedithio)-1-cyclopentene-1-carbaldehyde), 9 (4-benzyloxy-2-(1, 3-dioxa-2-cyclohexyl)-2-cyclopenten-1-one) and 20 (11a-hydroxy-13-oxaprostanoic acid) from (–)-quinic acid.

Intermediates recorded in this paper are also amenable to the preparation of all the prostanoids and biologically active five-membered and fused five-membered ring natural products [1].

One retrosynthetic analysis of the target molecules suggests that they can be constructed from an acyclic precursor **6** (3-benzyloxy-5, 5-(ethylenedithio)hexanedial by an intramolecular aldolisation-dehydration sequence. The requisite intermediate for the synthesis was the hydroxy ketone **1** (3,4-(1,1-cyclohexylenedioxy)-5-hydroxy-1-cyclohexanone) which was readily available in three steps and 75% overall yield from (-)-quinic acid [7].

Thus, treatment of 1 with ethanedithiol in the presence of boron trifluoride etherate in anhydrous chloroform furnished [8] 2(5,5-(ethylenedithio)-1,2,3-cyclo-

hexanetriol) in 95% yield, m.p. 129-130°, $[a]_D = -41°$ (c = 1.4, CH₃OH). The cyclohexylidene protecting group was needed for the preparation of the benzyl ether 4 (1-benzyloxy-2,3-(1,1-cyclohexylenedioxy)-5,5-(ethylenedithio)cyclohexane). Therefore the triol **2** was reconverted into its cyclohexylidene derivative **3** (2,3-(1,1-cyclohexylenedioxy)-5,5-(ethylenedithio)cyclohexanol (95%, m.p. 138-140°, $[a]_D = -44°$ (c = 1, CHCl₃)) by reaction with 1,1-dimethoxycyclohexane in the presence of a catalytic amount of sulfuric acid in *N*, *N*-dimethylformamide.

Benzylation of the free hydroxyl group in 3 using sodium hydride and benzyl bromide in N, N-dimethylformamide provided the crystalline 4 in 95% yield, m.p. $68-69^{\circ}$, $[a]_{p} = -50^{\circ}$ (c = 1.6, CHCl₃).

Acid hydrolysis of 4 using aq. acetic acid yielded the solid 5 (3-benzyloxy-5, 5-(ethylenedithio)-1, 2-cyclohexanediol in virtually quantitative yield, m.p. 135-136°, $[a]_{p} = -72^{\circ}$ (c = 1,05, CHCl₃).

Lead tetraacetate oxidation of the diol 5 in anhydrous toluene at room temperature afforded 6 in quantitative yield as a colourless oil, $[a]_{\rm p} = -11^{\circ}$ (c = 2,4, CHCl₃). As the dialdehyde is particularly unstable, it was directly cyclized under N₂ with a catalytic amount of 1N pyrrolidine acetate in anhydrous toluene at 0° overnight to give 7 in 95% yield. Acetalisation of 7 using 1,3 propanediol at room temperature in anhydrous toluene containing a catalytic amount of *p*-toluene-sulfonic acid produced 8 (4-benzyloxy-2-(1,3-dioxa-2-cyclohexyl)-1,1-(ethylene-dithio)-2-cyclopentene) in 80% yield, m.p. 70-71°, $[a]_{\rm p} = +86$ (c = 1.1, CHCl₃). The absolute configuration of the asymmetric C-atom of 8 was secured by X-ray crystallographic analysis [9] (*Figure*).

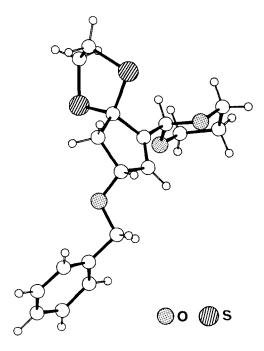
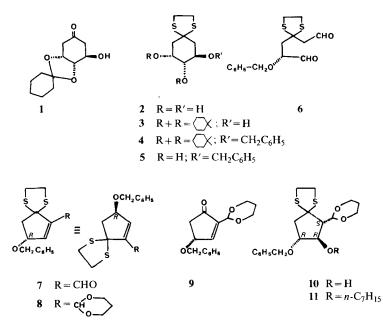


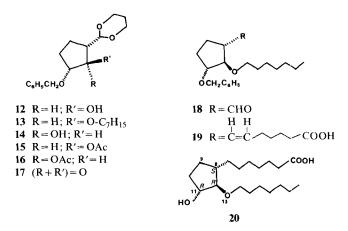
Figure. Three-dimensional view of $\mathbf{8}$ as determined by X-ray analysis



To demonstrate the versatility of 8, the latter was used as a relay compound for the synthesis of 9 (4-benzyloxy-2-(1, 3-dioxa-2-cyclohexyl)-2-cyclopenten-1-one) and 10 (2-benzyloxy-5-(1, 3-dioxa-2-cyclohexyl)-4, 4-(ethylenedithio) cyclopentanol).

The preparation of 9 was accomplished in straightforward fashion by treating 8 with phenylselenic anhydride [10] and propylene oxide to give 9 (68%), m.p. 42-43°, $[a]_{p} = +42^{\circ}$ (c = 1, CHCl₃).

The synthesis of 10 involved the hydroboration of 8 with diborane followed by alkaline hydrogen peroxide oxidation to yield the desired compound 10 (35%, $[a]_{D} = -40^{\circ}$ (c = 1.3, CHCl₃)). Alkylation of the latter using heptyl iodide, sodium



hydride in N, N-dimethylformamide furnished in 80% yield, the syrupy product 11 (1-benzyloxy-3-(1, 3-dioxa-2-cyclohexyl)-4, 4-(ethylenedithio)-2-heptyloxycyclopentane). It was expected that the hydroboration of 8 would proceed in an anti-Markovnikov fashion and the procedure would provide 10 with correct absolute configuration at the three asymmetric centres for the synthesis of 11a-hydroxy 13-oxaprostanoic acid 20. Raney-nickel desulfurisation of 10 and 11 gave 12 (2*a*-benzyloxy-5*a*-(1, 3-dioxa-2-cyclohexyl)cyclopentanol), $[a]_{D} = +5^{\circ}$ (*c*=0.75, CHCl₃), and 13 (1*a*-benzyloxy-3*a*-(1, 3-dioxa-2-cyclohexyl)-2 β -heptyloxycyclopentane), $[a]_{p} = +5^{\circ}$ (c = 1, CHCl₃) in yields of 70% and 75%, respectively. In order to corroborate further the stereochemical assignment of 12 and hence 20, the β alcohol 12 was oxidized with dipyridine-chromium(VI) oxide [11] to the corresponding keto derivative 17 (2-benzyloxy-5-(1, 3-dioxa-2-cyclohexyl)-1-cyclopentanone). Sodium borohydride reduction of the latter produced a mixture of the epimeric alcohols 12 and 14 (2-benzyloxy-5-(1, 3-dioxa-2-cyclohexyl)-la-cyclopentanol) in a ratio of 13:87, which were separated and readily converted into their acetates 15 (1 β -acetoxy-2-benzyloxy-5-(1, 3-dioxa-2-cyclohexyl)cyclopentane) and 16 (1a-acetoxy-2-benzyloxy-5-(1, 3-dioxa-2-cyclohexyl)cyclopentane).

Differentiation between the two structures **15** and **16** was possible on the basis of their ¹H-NMR. parameters by comparison with appropriate literature values [12] [13]. [¹H-NMR. (80 MHz, benzene- d_6): (of *trans*-acetate **15**): 2.4 (*m*, $J_{1,5}$ =4.5, 1 H, H–C(5)); 5.75 (*qa*, $J_{1,2}$ =4.5, 1 H, H–C(1)); 3.9 (*m*, 1 H, H–C(2))]; (of *cis*-acetate **16**): 2.25 (*m*, $J_{1,5}$ =3.5 Hz, 1 H, H–C(5)); 5.9 (*t*, $J_{1,2}$ =3.5, 1 H, H–C(1)); 3.6 (*m*, 1 H, H–C(2)).

Moreover, the ¹³C-NMR. data of the parent alcohols '*trans*'-12 and '*cis*'-14 were in excellent agreement with the assigned structures. The chemical shifts of the C-atoms C(1), C(2), C(5) of the '*cis*'-isomer 14 appeared at higher field [13] [14] than those of the '*trans*'-isomer 12. [¹³C-NMR. (CDCl₃); (of '*trans*'-12): 49.1 (C(5)); 79.7 (C(1)); 85 (C(2)); (of '*cis*'-14): 46.3 (C(5)); 71.5 (C(1)); 81.4 (C(2))].

The attachement of the carboxylic acid side chain to the aldehyde **18** (readily obtained by acid hydrolysis of the acetal function in **13** (92%)) relied on the *Wittig* condensation and proceeded smoothly. Thus, treatment of **18** (3-benzyloxy-2-heptyloxy-cyclopentanecarbaldehyde) with the ylide derived from 5-carboxypentyl-triphenylphosphonium bromide gave the oily unsaturated product **19** (11*a*-hydroxy-6, 7-dehydro-13-oxaprostanoic acid) (55%), $[a]_D = -45^\circ$ (c = 0.35, CHCl₃). Subsequent catalytic hydrogenation of the double bond with concomitant hydrogenolysis of the benzyl group in **19** was achieved in 90% yield affording the target 11*a*-hydroxy-13-oxaprostanoic acid **20**, $[a]_D = +19^\circ$ (c = 0.75, CHCl₃).

While the focus of this work has been the preparation of the synthons 7, 8, 9 and 20 11a-hydroxy 13-oxaprostanoic acid), it is clear that all prostanoids and their congeners and other cyclopentanoid natural products are accessible through applications of the present strategy¹).

¹) Satisfactory IR. spectra, ¹H- and ¹³C-NMR. spectra, mass spectra, and analytical data were obtained on purified, chromatographically homogeneous samples of all intermediates here.

REFERENCES

- B. M. Trost, 'Stereoselective synthesis of natural products', W. Barmann & E. Winterfeldt, Excepta Medica, Amsterdam, Oxford, 1979, 106.
- [2] J. S. Bindra & R. Bindra, 'Prostaglandin synthesis', Academic Press, New York 1977.
- [3] P. Crabbé, 'Prostaglandin Research', Academic Press, New York 1977.
- [4] R.M. Scarborough, jr., B.H. Toder & A.B. Smith, J. Am. Chem. Soc. 102, 3904 (1980).
- [5] S. Köksal, P. Raddatz & E. Winterfeldt, Angew. Chem. Int. Ed. Engl. 19, 472 (1980).
- [6] S.J. Branca & A.B. Smith, III, J. Am. Chem. Soc. 100, 7767 (1978).
- [7] J. Cléophax, S.D. Géro, J. Leboul, M. Akhtar, J.E.G. Barnett & C.J. Pearce, J. Am. Chem. Soc. 98, 7110 (1976).
- [8] D. H. R. Barton, S. D. Géro & C. D. Maycock, J.C.S. Chem. Comm. 1980, 1089.
- [9] J. C. Barrière, A. Chiaroni, J. Cléophax, S. D. Géro & C. Riche, Acta Crystallogr. (in preparation).
- [10] D. H. R. Barton, N.J. Cussans & S. V. Ley, J.C.S. Chem. Commun. 1978, 393.
- [11] J.C. Collins, W.W. Hess & F.J. Frank, Tetrahedron Letters 1968, 3363.
- [12] R. Hanselger & P. De Clercq, Org. Mag. Res. 13, 376 (1980).
- [13] D.L. Venton, Steven E. Enke & G.L. LeBreton, J. Med. Chem. 22, 824 (1979).
- [14] Manfred Christl, Hans J. Reich & John D. Roberts, J. Am. Chem. Soc. 93, 3463 (1971).